Synthetic Approaches towards New Bisformazans and Bisverdazyls

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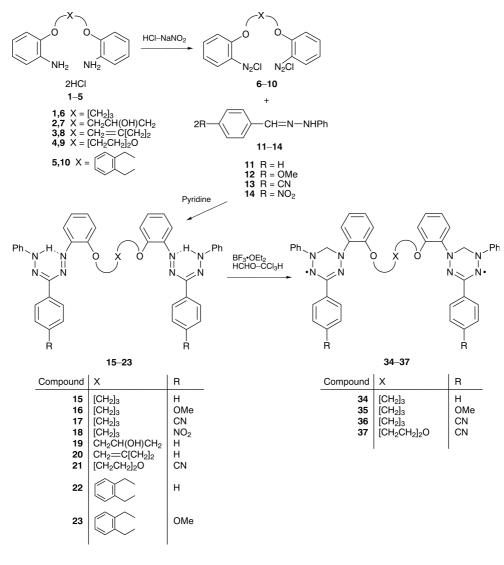
The new bisformazans **15–23**, **28** and **29** are prepared *via* three routes and some are converted into the corresponding bisverdazyls **34–37** which represents a new class of stable diradicals.

Much attention has been directed towards the chemistry of formazans owing to their diverse applications in many fields.^{1,5,8,27} Moreover, verdazyls discovered by Kuhn and Trischman^{17,18} represent a class of extraordinary stable free radicals with many interesting applications.²⁶ The present investigation describes three synthetic routes for the synthesis of the bisformazans **15–23**, **28** and **29** (Schemes 1 and 2) and the conversion of some to the corresponding bisverdazyls **34–37**.

Thus, coupling of the bisdiazonium salts 6-10 (prepared from the corresponding bisamines 1-5) with the appropriate araldehyde phenylhydrazones 11-14 in pyridine gave

the corresponding bisformazans 15–23 in 56–64% yield. Following a recent recommended method using phase transfer catalytic synthesis of formazans²⁹ the yields of compounds 17, 19 and 21 were increased to 69-70% when the coupling reaction was carried out in dichloromethane containing either Na₂CO₃ or NaOH and tetrabutyl-ammonium iodide as PTC.

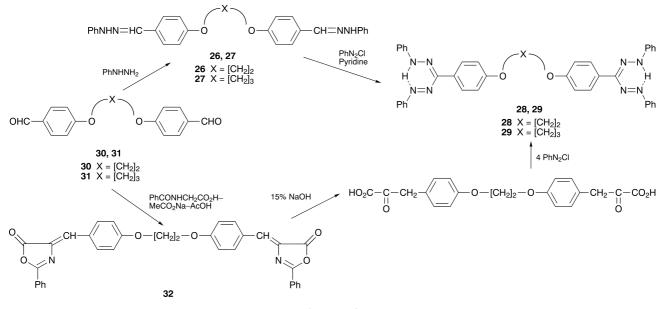
On the other hand, the bisformazans **28** and **29** were prepared from the biscarboxaldehyde derivatives **30**, **31** by converting them into the corresponding bisphenylhydrazones followed by coupling with benzenediazonium chloride in pyridine. Alternatively, compound **28** was prepared



Scheme 1

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Scheme 2

from 1,2-bis(*p*-formylphenoxy)ethane **30** by converting it first to the bisarylpyruvic acid **33** (*via* the intermediate bisoxazolone **32**) followed by coupling with benzene-diazonium chloride.

Finally compounds 15–17 and 21 were converted into the interesting new bisverdazyls 34-37 upon treatment with paraformaldehyde and boron trifluoride diethyl etherate in chloroform, followed by aqueous formaldehyde and then aqueous sodium hydroxide as reported.³³

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Techniques used: ¹H and ¹³C NMR, MS, ESR

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